# Thermal Management Systems and Waste Heat Recycling by Thermoelectric Generators-An Overview 

Sadeq Hooshmand Zaferani ${ }^{1}{ }^{(D}$, Mehdi Jafarian ${ }^{2}$, Daryoosh Vashaee ${ }^{3,4}{ }^{(D}$ and Reza Ghomashchi ${ }^{1,5,6, *}$<br>1 School of Mechanical Engineering, University of Adelaide, Adelaide 5005, Australia; sadeq.hooshmandzaferani@adelaide.edu.au<br>2 Centre for Energy Technology, School of Mechanical Engineering, University of Adelaide, Adelaide 5005, Australia; mehdi.jafarian@adelaide.edu.au<br>3 Department of Electrical and Computer Engineering, North Carolina State University, Raleigh, NC 27606, USA; dvashae@ncsu.edu<br>4 Department of Materials Science and Engineering, North Carolina State University, Raleigh, NC 27606, USA<br>5 ARC Research Hub for Graphene Enabled Industry Transformation, University of Adelaide, Adelaide 5005, Australia<br>6 Institute for Photonics and Advanced Sensing, University of Adelaide, Adelaide 5005, Australia<br>* Correspondence: reza.ghomashchi@adelaide.edu.au

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#### Abstract

With the fast evolution in greenhouse gas (GHG) emissions (e.g., $\mathrm{CO}_{2}, \mathrm{~N}_{2} \mathrm{O}$ ) caused by fossil fuel combustion and global warming, climate change has been identified as a critical threat to the sustainable development of human society, public health, and the environment. To reduce GHG emissions, besides minimizing waste heat production at the source, an integrated approach should be adopted for waste heat management, namely, waste heat collection and recycling. One solution to enable waste heat capture and conversion into useful energy forms (e.g., electricity) is employing solid-state energy converters, such as thermoelectric generators (TEGs). The simplicity of thermoelectric generators enables them to be applied in various industries, specifically those that generate heat as the primary waste product at a temperature of several hundred degrees. Nevertheless, thermoelectric generators can be used over a broad range of temperatures for various applications; for example, at low temperatures for human body heat harvesting, at mid-temperature for automobile exhaust recovery systems, and at high temperatures for cement industries, concentrated solar heat exchangers, or NASA exploration rovers. We present the trends in the development of thermoelectric devices used for thermal management and waste heat recovery. In addition, a brief account is presented on the scientific development of TE materials with the various approaches implemented to improve the conversion efficiency of thermoelectric compounds through manipulation of Figure of Merit, a unitless factor indicative of TE conversion efficiency. Finally, as a case study, work on waste heat recovery from rotary cement kiln reactors is evaluated and discussed.


Keywords: energy conversion; waste heat recovery; thermoelectric generators; thermal management

## 1. Introduction

Global warming and the resulting climate change can be traced back to human activities, industrial processes, and the resultant greenhouse gases (GHGs) emissions [1,2]. The primary source of GHGs emissions is the application of fossil fuels for energy production. Therefore, one of the ways to reduce the emission of GHGs is to consume less fossil fuel. However, this is only feasible if the energy is generated and used efficiently, i.e., having the minimum energy loss and recycling the energy loss, which usually comes as waste heat. According to a recent study conducted by the Lawrence Livermore National Laboratory [3], more than $68 \%$ of the generated energy is lost as waste heat, Figure 1. However, amongst the energy generation sectors, the majority of the rejected energy (waste heat) in the United States belongs to electricity generation ( $24.2 \%$ ), in comparison with the other sectors, in-
cluding transportation (22.3\%), industrial (13.5\%), residential (4.17\%), and commercial (3.29\%), as illustrated in Figure 1.

Estimated U.S. Energy Consumption in 2019: 100.2 Quads


Figure 1. Energy flow diagram of the US energy production and consumption. Source: Lawrence Livermore National Laboratory [3].

It may take many years until renewable energies reach a position where they can fully replace fossil fuels. Consequently, other intermediate technologies are required to reduce carbon output (e.g., $\mathrm{CO}_{2}$ ) and increase the efficiency of the existing energy-generating technologies during this transition period. In this regard, Figure 2 illustrates the electricity generation from fuel energies (Figure 2a) and renewable energy sources (Figure 2b) in the US, which are projected up to 2050. It is anticipated that a greater role for renewable energy sources will satisfy future energy demand, as clearly demonstrated in Figure 3.
(a) Electricity generation from selected fuels (AEO2020 Reference case)
billion kilowatthours


Renewable electricity generation, including end use (AEO2020 Reference case)
billion kilowatthours


Figure 2. Electricity generation from selected energy resources, (a) fuel energies and (b) clean energies [4].


Figure 3. Application of renewable energies in the future [4].
The forecast in Figure 3 is based on the trend of rapid deployment of clean energy technologies, which is encouraging. Such growth in the renewable energy industry is expected to show its impact on the projected $\mathrm{CO}_{2}$ emission in the electric power sector, as exhibited in Figure 4. As shown in this figure, the average $\mathrm{CO}_{2}$ emissions decrease with time.


Figure 4. Energy-related $\mathrm{CO}_{2}$ emissions by the energy sector [4].
The reduction in $\mathrm{CO}_{2}$ emission is related to energy harvesting from the environment by reliable sources, which has now manifested as a worldwide concern. Currently, the relatively low-performance efficiency of the fuel-based engines, industrial and power generation processes leads to a significant fraction of the energy of the fuels being lost as heat. For example, the thermal efficiency of the state-of-the-art petrol engines is $\sim 20 \%$, gasoline engines are $\sim 40 \%$, cement kilns are $\sim 50 \%$, and Rankine cycles for power generation are some $35 \%[4,5]$. Therefore, if the loss of valuable energy to waste heat is reduced through more efficient processes, or the waste heat is converted into more useful energy such as electricity, the environmental impact of energy production will be mitigated.

One solution to capturing waste heat and converting it into electricity is utilizing solid-state energy converters, such as thermoelectric generators (TEGs). In line with this encouraging trend, the application of thermoelectric generators is considered as the
potential primary candidate to capture waste heat, for example, in automobile exhaust recovery systems [5-7], or harvesting the heating energy dissipating from current renewable energy sources (e.g., concentrated solar heat exchangers) [8]. Therefore, the central theme of this paper is to discuss the advances in thermoelectricity and its applications in energy recovery to highlight the benefits of thermoelectric generators in converting waste heat to electricity without any taxing on the environment.

## 2. History of Thermoelectricity

The concept of thermoelectricity (TE) as the direct conversion of temperature gradient $(\nabla \mathrm{T})$ into electricity was instigated by the German-Estonian physicist Thomas Johann Seebeck in the years between 1821 and 1831 [9,10]. The Seebeck coefficient (S, V/K) was estimated based on the generated voltage $(\Delta \mathrm{V})$ against an imposed temperature difference $(\Delta T)$ [6]-Equation (1),

$$
\begin{equation*}
S=-\frac{\Delta V}{\Delta T} \tag{1}
\end{equation*}
$$

Further contributions in this field are due to Gustav Magnus, who declared the Seebeck voltage is not dependent on the temperature distribution along the intermediate metals connecting the p-n junctions [11]. Later on, the Peltier effect was introduced to show that the heating or cooling is generated by passing an electric current through two dissimilar materials [9,10]. Furthermore, William Thomson defined the relationship between the Seebeck and Peltier effects [10], and then Edmund Altenkirch calculated the maximum efficiency of the thermoelectric generators [9]. The modern theory of thermoelectricity in materials, however, is due to Abram Fedorovich Ioffe [9], who introduced a dimensionless factor named "figure of merit (zT)". It is a metric for evaluating the thermoelectric behavior (performance) of thermoelectric compounds (materials);

$$
\begin{equation*}
\mathrm{zT}=\frac{\mathrm{S}^{2} \sigma}{\mathrm{k}_{\mathrm{e}}+\mathrm{k}_{\mathrm{l}}} \mathrm{~T} \tag{2}
\end{equation*}
$$

where $S$ is the Seebeck coefficient $\left(\mu \mathrm{VK}^{-1}\right)$, $\sigma$ is the electrical conductivity $\left(\Omega^{-1} \mathrm{~m}^{-1}\right)$, T is the temperature $(\mathrm{K})$, and $\kappa_{\mathrm{e}}$ and $\kappa_{1}$ are electronic and lattice thermal conductivities ( $\mathrm{Wm}^{-1} \mathrm{~K}^{-1}$ ), respectively [12]. The following chart (Figure 5) summarizes the historical trend in thermoelectric development.

Following the introduction of the fundamentals on thermoelectricity during the 19th and 20th centuries, the 21st century is witnessing the application of the TE fundamentals in developing advanced technology through the fabrication of thermoelectric generators and their subsequent use in introducing intelligent devices with immense effects on the well-being of our society.


Figure 5. Historical trends in thermoelectric development [13-15].
Figure 6 provides some examples of the recent applications of thermoelectric generators. According to this figure [16-24], the exploration of new TE materials has extended the application of the TEGs from low-temperature range, such as room temperature for capturing human body heat $[18,20]$, to medium temperature for recovering car exhaust heat [21], and then they have been functionalized at high temperatures for NASA emissions to empower discovery rovers [23,24].


Figure 6. Recent decades: 21st century: (a) self-doped conjugated polyelectrolytes (CPEs) has been evaluated as a promising class of conductive organic for organic thermoelectrics (adapted from [16] with permission), (b) photograph of a prototype for a compact and flexible TEG (CF-TEG) using ultra-fine chip mounting technique. It is made of 84 p-n pairs fabricated on a $10 \times 10 \mathrm{~mm}^{2}$ flexible substrate. It produced a maximum output voltage and power density of 2.4 V and $185 \mathrm{mWcm}^{-2}$, respectively, at $\Delta \mathrm{T}=150 \mathrm{~K}$ (adapted from [17] with permission), (c) photograph of a flexible TEG prototype. It is fabricated by a locknit spacer fabric as the substrate and yarns are coated with waterborne polyurethane/carbon nanotube thermoelectric composites as legs (thermocouples), (adapted from [18] with permission), (d) body-heat harvester made with cotton. A vapor-deposited strip of cotton fabric-coated with p-doped poly(3,4-ethylenedioxythiophene) [19]. Photo source: IEEE spectrum [20], (e) an automotive thermoelectric generator (ATEG) is built and tested experimentally on the exhaust system of the engine cell (adapted from [21] with permission), (f) schematic illustration of the solar-thermal-electric conversion system (STECS) constructed by a movable charging solar-thermal storage and TE conversion subsystems. The magnet installed at the bottom of the solar-thermal storage subsystem is to make movable charging of the molten salts inside the container (adapted from [22] with permission), (g) installation of thermoelectric generator systems on the Cassini spacecraft, photo credit: NASA [23], and (h) NASA's Mars 2020 rover is powered based on thermoelectric power generation [24].

## 3. Thermoelectric Materials and Designs

The efficiency of thermoelectric generators is very much dependent on the conversion efficiency of the constituent thermoelectric (TE) compounds [25-29]. Therefore, the introduction of advanced thermoelectric materials is crucial in increasing the waste heatelectricity conversion rate. Today, the central theme of research in different laboratories worldwide is to explore and improve thermoelectric materials.

The importance of $z T$ as the main material-dependent factor is evident when the maximum efficiency $\left(\eta_{\max }\right)$ of the thermoelectric (TE) generators ( p and n junction) is calculated as;

$$
\begin{equation*}
\eta_{\max }=\frac{T_{H}-T_{C}}{T_{H}} \times \frac{\sqrt{1+Z T_{M}}-1}{\sqrt{1+Z T_{M}}+\frac{T_{c}}{T_{H}}} \tag{3}
\end{equation*}
$$

where $T_{H}$ and $T_{C}$ are the temperatures of the hot and cold surfaces, respectively, and $Z T_{M}$ is the figure of merit at the average temperature $\left(T_{M}\right)$ of the hot and cold junctions [30]. As evident in Equation (3), a large thermal gradient ( $\Delta T=T_{H}-T_{C}$ ) can increase the TE efficiency. However, a large thermal gradient may also impose thermal stresses that may compromise the mechanical integrity of the TE compounds. Therefore, in addition to the appropriate thermoelectric characteristics, the TE materials should have thermal and mechanical stabilities to function under harsh conditions.

There is a range of thermoelectric materials classified as clathrates [31-33], skutterudites [34-37], chalcogenides [38-40], and half-Heuslers [41-48]. However, most studies are concentrated on microstructural manipulation to improve the thermoelectric parameters $(S, \sigma, \kappa)$ at various temperatures. In this regard, the stoichiometry of the p and n-type TE compounds is the critical variable. For example, Figure 7 shows $z T$ values for ther-
moelectric products at different temperatures. As illustrated in this figure, different TE compounds with various $z T$ values confirm their dependency on the constituent elements.


Figure 7. (a) Stem chart, which classifies the TE materials based on their optimum operating temperatures in three main intervals: low (Room temperature to 600 K ), mid ( 600 to 850 K ), and high ( 850 K and more) [49], (b) a list of TE compounds, and a TEG operating at room temperature, (c) illustrates the TE materials with the optimized application at medium temperatures, and (d) an application of TEGs in converting car exhaust heat into electricity [5,50], (e) represents the TE compounds with optimum performance at high temperatures, and (f) two examples of their applications by NASA [51], and Solar TEGs [52] (adapted from [8], with permission).

The search for advanced TE materials reveals the necessity of designing products containing nontoxic/eco-friendly elements, sustainable at harsh conditions and high temperatures. Based on the dimensionless figure of merit (Equation (2)), there should be a correlation between the thermoelectric parameters to make an efficient TE product with reasonable $z T$. Therefore, the effective factors, including the Seebeck (S), and electrical ( $\sigma$ ), and thermal ( $\kappa$ ) conductivities, must be optimized, considering the thermal stability of the materials. The TE factors of $S$ and $\sigma$ are mainly related to the electronic band structure
parameters, usually calculated from the first principles. However, thermal conductivities can be modified via microstructural manipulation to increase the heat carriers scattering and reduce thermal conductivities. Significant efforts by the TE community over the last few decades have been concentrated on various techniques to improve thermoelectric properties, based on modifying $S, \sigma$, and $\kappa$ [53-57], such as nanostructuring [58-61]. For instance, Figure 8 illustrates the impact of microstructure through changes in particle morphology on the thermoelectric properties, including electrical conductivity, Seebeck coefficients, lattice thermal conductivity, and power factor. The reported impact of the microstructure on thermoelectric factors was discussed based on the modification in carrier concentration, carrier mobility, and effective mass [62]. The readers are referred to this work [62] for more information since the further discussion is outside the scope of the current paper.


Figure 8. SEM micrographs of the sintered (a) $\mathrm{Bi}_{2} \mathrm{Te}_{3}$ nanoparticles (BT-1), (b) $\mathrm{Bi}_{2} \mathrm{Te}_{3}$ thick nanoplates (BT-2), and (c) ultrathin $\mathrm{Bi}_{2} \mathrm{Te}_{3}$ nanosheets (BT-3). Temperature dependence of (d) electrical conductivity, and Seebeck coefficient, (e) lattice thermal conductivity, and power factor (adapted from [62] with permission).

Another approach for manipulation of microstructure through reinforcing nanoparticles such as graphene nanoplates (GNPs) has been investigated in previous works of this group $[28,63]$ for two thermoelectric nanocomposites, namely MnTe-GNPs and CoVSnGNPs (Figure 9). The results revealed an improved modification in the thermoelectric parameters, mainly in the Seebeck coefficients ( $27 \%$ increase) and thermal conductivity ( $34 \%$ decrease) for the samples reinforced with $0.25 \mathrm{wt} . \%$ GNPs in comparison to the pristine MnTe samples at the temperature of 600 K . Moreover, dispersion and precipitation of GNPs in the CoVSn multiphase microstructure influenced the thermoelectric factors by reducing the thermal conductivity and increasing the Seebeck coefficient, leading to the enhancement of the thermoelectric figure of merit. The change in the thermoelectric factor
was attributed to the phonon scattering at the grain boundaries beside carrier filtering against the band bending (Schottky barriers) (see Figure 9d-f).


Figure 9. (a) Reinforcing MnTe by graphene nanoplates (GNPs) via powder metallurgy, (b) GNPs precipitation in the matrix (adapted from [63] with permission), (c) GNPs precipitation at microstructural barriers of CoVSn compound (adapted from [28] with permission), (d) schematic band alignment of Schottky contact for graphene/n-type semiconductor (adapted from [28] with permission), (e) schematic band alignment of Schottky contact for graphene/p-type semiconductor (adapted from [28] with permission), (f) schematic diagrams of hole trap, hole barrier, electron trap, and electron barrier generated at grain boundaries (adapted from [28] with permission).

Many more strategies have been studied to enhance the performance of thermoelectric compounds. The examples would be materials selection (as described in Figure 7) [64-66], sintering method (e.g., pondermotive force in microwave sintering enhances the diffusion of mobile ionic species and results in accelerating the solid-state reaction by increasing the collision probability) [67], band engineering (e.g., modulation doping, resonance level, and band convergence) [68-74], carrier pocket engineering [75-77], complex structures [78,79], carrier energy filtering [80,81], creation of resonant energy levels close to the band edges [70], low dimensional structures [82,83], magnetic interaction (e.g., carrier trap-
ping and magnon (spin wave) excitations) [84,85], and lowering the thermal conductivity (e.g., phonon scattering) [58,86-90].

## 4. Waste Heat Sources

Over the past few years, assessing the primary energy-user sectors reveals that a large amount of thermal energy is wasted (Figure 10).


Figure 10. Schematic illustration of several sources of waste thermal energy and waste heat recycling by thermoelectric generators.

According to the estimation [18], several segments have contributed to wasting thermal energies, from industrial to domestic applications, as described in Figure 10, which can be potentially equipped with thermoelectric generators for converting (recycling) the waste heat into electricity. As a case study and to further justify the rationale and economic benefits behind TEG application in the industrial sectors, the preliminary studies on waste heat recycling from rotary kiln surfaces are discussed and analyzed in the subsequent sections.

## 5. Case Study: Waste Heat Recycling from Cement Rotary Kilns

Cement subsectors are considered among the most energy-intensive industries, which absorb about $12-15 \%$ of the total energy consumed by the industry [91]. This industry is perhaps one of the most polluting industrial sectors, producing about $7 \%$ of the manmade carbon dioxide emissions each year [92]. In theory, approximately 1674-1799 kJ energy with no heat loss is required to produce one kilogram of cement clinker [93]. However, in practice, the required energy to manufacture one kilogram of cement clinker is reported in the range of $2750-4500 \mathrm{~kJ}$ [93]. Also, to produce one tonne of cement, average electrical energy of $110-120 \mathrm{kWh}$ is required [93]. The cement rotary kiln is the heart of the manufacturing stage for large-scale industrial cement production. In this process, basic raw materials are continuously heated along the kiln length for dehydration, preheating, and calcination with a significant amount of parasitic heat lost through the rotary kiln shell, which is needed to lower the temperature of the shell relative to that of the reactor inside, which can result in temperature to $\sim 1450^{\circ} \mathrm{C}$ over the clinkering part of the reactor. The heat loss from the outside surface of the shell enables keeping the operating temperature of the shell structure within a reasonable range and, hence, increasing the lifetime of the shell. This also implies the significance of developing appropriate heat recovery systems from the shell, enabling the conversion of the lost heat into energy without changing operating conditions. Commensurate with this, the calculations in a case study [94] showed that the mean daily heat loss via the calcination zone of a kiln shell with a production capacity of 2500 tons/day
is about 140 GJ , attributing to radiation and convection with fractions of $\sim 54 \%$ and $\sim 46 \%$, respectively. The industrial cement sectors have been considering recycling waste heat to achieve higher rates of energy conservation and, thus, financial and environmental advantages [91,95-97]. Nevertheless, as mentioned above, any employed strategies for heat recovery need to avoid/minimize changes in the temperature distribution within the kiln shell. As discussed above, TEGs can be potentially used for waste heat recovery from the rotary kilns and turn the heat directly into electrical energy [96,97], given that the TEGs and their arrangement around the kiln shell can be purpose-designed to minimize the change in the temperature distribution in the shell, while maximizing their output.

A recent study by Mirhosseini et al. [95] assessed the application of TEG for heat recycling in an Aalborg Portland rotary cement kiln. This investigation proposed applying a local thermal annular absorber integrated with a thermoelectric system to optimize waste heat recovery from the kiln. The absorber collects the heat from the kiln surface, transferred via convection and radiation. Figure 11a shows a view of a cement rotary kiln in the Aalborg Portland cement factory. A numerical simulation was undertaken to evaluate heat loss from the rotary cement kiln. For this purpose, the body temperature of the absorber was determined to report the temperature distribution (Figure 11b). Through the calculations, the surface temperature and free stream air temperature were set to $500^{\circ} \mathrm{C}$ and $5{ }^{\circ} \mathrm{C}$, respectively, for the cement kiln shown schematically in Figure 11 b (inset). The cement kiln diameter was set at 3.6 m , and the inner diameter of the absorber, made of an aluminum alloy sheet, was considered 4 m . Three heat transfer mechanisms, including free convection, forced convection due to kiln rotation, and radiation, were applied to calculate the heat loss from the external surface of the kiln, as shown in Figure 11c. The simulation calculation domain was carried out with a two-dimensional (2D) unsteady solution, as illustrated in Figure 11d. In this simulation, the one-dimensional thermal equivalent electrical circuit shown in Figure 11e and the electrical resistance networks presented in Figure 11f were considered to calculate the electrical power output for the thermoelectric unit installed on the absorber, see Figure 11g.

The optimization of the simulated TEG was undertaken via a mathematical model based on the finite element method (FEM). To design a thermoelectric recovery generator, two different thermoelectric materials, including bismuth telluride $\left(\mathrm{Bi}_{2} \mathrm{Te}_{3}\right)$ and $\beta$-phase zinc antimonide $\left(\mathrm{Zn}_{4} \mathrm{Sb}_{3}\right)$, were used to calculate the electrical power output and TEG performance by considering the financial aspects (e.g., investment cost and production benefits).

Table 1 presents the results of this work [95] according to various parameters, such as the TEG fill factor, leg length of thermocouples, TEG matched power output, total investment cost, cost per power ratio, and payback period. Based on the project outcome, if the cement kiln operates continuously for a year ( 8760 h ), considering the electricity price for household consumption in Denmark ( $\$ 0.35 / \mathrm{kWh}$ ), the optimized TEG system fabricated by $\mathrm{Bi}_{2} \mathrm{Te}_{3}$ - based thermoelectric materials should take about 8.3 years to repay the investment cost. However, the optimized TEG made by $\mathrm{Zn}_{4} \mathrm{Sb}_{3}$ - based thermoelectric compounds will pay back the investment cost in about 3.5 years. Subsequently, the recovery systems are profitable after these periods.


Figure 11. (a) A view of a cement rotary kiln in the Aalborg Portland cement factory, Aalborg, Denmark. Length 163 m , outer diameter: 3.6 m (adapted from [95] with permission), (b) temperature distribution along the kiln surface with a measurement interval of 2 m , inset: schematic illustration of the studied annular absorber (inset). The free stream air temperature ( $\mathrm{T} \infty$ ) and corresponding velocity ( $\mathrm{U} \infty$,) were considered as $5^{\circ} \mathrm{C}$ and $5.86 \mathrm{~m} / \mathrm{s}$, respectively. The kiln rotational speed: 5 rpm (clockwise) (adapted from [95] with permission), (c) local heat loss through the kiln surface based on three heat transfer mechanisms, including free convection, forced convection, and radiation (adapted from [95] with permission), (d) schematic illustration of the 2D computational domain (adapted from [95] with permission), (e) one dimensional thermal equivalent electrical circuit (adapted from [95] with permission), (f) electrical resistance networks (adapted from [95] with permission), (g) schematic illustration of the thermoelectric unit on the panel (adapted from [95] with permission).

Table 1. Optimum design characteristics for the TEG systems fabricated by $\mathrm{Bi}_{2} \mathrm{Te}_{3}$ and $\mathrm{Zn}_{4} \mathrm{Sb}_{3}$-based thermoelectric compounds [95].

|  | Optimum Fill Factor | Optimum <br> Leg Length (mm) | Optimum Power Output (W) | Optimum Power Output (W/m ${ }^{2}$ ) | Cost (\$US) | Cost per Power (\$US/W) | Payback Period (year) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Bi}_{2} \mathrm{Te}_{3}$ | 0.01 | 4 | 825.0 | 64.8 | 31,517 | 38.2 | 12.46 |
|  | 0.05 | 8.8 | 1511.9 | 118.8 | 38,465 | 25.44 | 8.30 |
|  | 0.1 | 12.4 | 1573.5 | 123.7 | 51,106 | 32.48 | 10.59 |
|  | 0.2 | 16.7 | 1430.4 | 112.4 | 83,007 | 58.03 | 18.93 |
|  | 0.5 | 26 | 1110.7 | 87.3 | 223,460 | 201.2 | 65.62 |
|  | 0.7 | 31 | 992.299 | 77.9 | 347,320 | 350.012 | 114.16 |
|  | 0.9 | 36.1 | 922.405 | 72.5 | 499,240 | 541.24 | 176.53 |
| $\mathrm{Zn}_{4} \mathrm{Sb}_{3}$ | 0.01 | 1.6 | 855.6 | 67.2 | 30,980 | 36.21 | 11.81 |
|  | 0.05 | 2.6 | 2558 | 201 | 32,677 | 12.77 | 4.17 |
|  | 0.1 | 4.2 | 3175.1 | 249.5 | 34,877 | 10.9845 | 3.58 |
|  | 0.2 | 7.5 | 3564 | 280.1 | 39,530 | 11.09 | 3.62 |
|  | 0.5 | 16 | 3818.8 | 300.1 | 55,136 | 14.44 | 4.71 |
|  | 0.7 | 21 | 3861 | 303.5 | 66,743 | 17.29 | 5.64 |
|  | 0.9 | 25.2 | 3867.9 | 303.9 | 79,000 | 20.424 | 6.66 |

In another study [98], a thermoelectric waste heat recycling system was applied to a rotary cement kiln. The surface temperature on the rotary cement kiln (length: 76 m , outer diameter: 4.8 m ) was set to $300^{\circ} \mathrm{C}$ with emitting heat flux equal to $3500 \sim 4000 \mathrm{~W} / \mathrm{m}^{2}$ (Figure 12a). The designed waste heat recycling system was constructed with a heat collection plate, TEG modules, and a water-cooled plate (Figure 12b). The system was installed 0.1 m away from the kiln surface. This system included 16 TEG units installed on a $2 \times 1 \mathrm{~m}^{2}$ steel frame to harvest the waste heat (Figure 12b). Figure 12c shows the output voltage as a function of current based on the temperature difference ( $\Delta T=T_{h}-T_{c}$ ) of $120^{\circ} \mathrm{C}$, in which $T_{h}=160^{\circ} \mathrm{C}$ is the temperature of the heat collector (the aluminum sheet) and $T_{\mathcal{C}}=40^{\circ} \mathrm{C}$ stands for the temperature of cooling water at the site. The impedance matching for $\Delta T=120^{\circ} \mathrm{C}$ in the system was measured against various external loads and a power electrical resistor array (series-connected) from 0.3 to $12 \Omega$. At operational conditions (i.e., kiln surface temperature: $280 \sim 320^{\circ} \mathrm{C}$ ), the emitted waste heat was absorbed by the heat collector via the heat transfer mechanisms of radiation and convection. In this process, the average temperature of the heat collector, installed 0.1 m away from the kiln surface, reached $160^{\circ} \mathrm{C}$ with a corresponding heat flux of $3500 \sim 3700 \mathrm{~W} / \mathrm{m}^{2}$. The power output generated by the system was used for indoor lighting.

According to the field-test results, this thermoelectric power generation system was reported to generate a maximum matched load power of 214.3 W (Figure 12e) at the site (close to the rotary kiln) and 169 W in the control room. Furthermore, as shown in Figure 12f, the generated electrical power was directly utilized to light up the indoor LED lamps ( 15 LED lamps $\times 10 \mathrm{~W}$ ) to contribute to the energy-saving systems [98].

Luo et al. have also [93] developed a mathematical model for estimating the performance of waste heat recycling by thermoelectric generators from a cement rotary kiln with a diameter of 4.8 m and a length of 72 m (see Figure 13).


Figure 12. (a) working cement kiln, outer diameter: 4.8 m and length: 76 m (adapted from [98] with permission), (b) thermoelectric power generation system containing 32 thermoelectric power generation units installed on a $2 \times 1 \mathrm{~m}^{2}$ steel frame (adapted from [98] with permission), (c) output voltage as a function of current (adapted from [98] with permission), (d) power output as a function of load resistance (adapted from [98] with permission), (e) generated power with a matched load power of 214.3 W at the site (adapted from [98] with permission), and (f) generated power with a matched load power of 214.3 W in the control room to light up the indoor LED lamps ( 15 LED lamps $\times 10 \mathrm{~W}$ ) (adapted from [98] with permission).


Figure 13. Configuration of a thermoelectric waste heat recycling system for a Portland cement rotary kiln, length: 72 m , diameter: 4.8 m , (adapted from [93]).

The results showed that a thermoelectric generator fabricated based on $\mathrm{Bi}_{2} \mathrm{Te}_{3}-\mathrm{PbTe}$ hybrid thermoelectric materials can generate about 211 kW electrical power to save 3283 kW thermal energy. Furthermore, the kiln integrated with this TEG system was estimated to be approximately $33 \%$ of the wasted energy compared to a kiln with no thermoelectric recovery system.

Another example is the application of TEG for waste heat recovery systems emitted from the surface of the rotary kiln performed for the Rudniki cement industrial plant [99]. In this work, the mentioned recovery system was designed for a rotary cement kiln with a length of 16-28 m. The studied TEG design generated a power output in the range of 1.46 kW to $\sim 11 \mathrm{~kW}$, corresponding to the received radiation heat of 10 kWh and 163 kWh , respectively, depending on the distance of the TEG system from the kiln surface. The
maximum energy generated by the designed TEG was reported as 163 kWh and 11 kW when the system was installed at a distance of 1 m from the rotary kiln surface. According to the discussion, since the cement industry has high energy consumption, resulting in waste heat and generated GHGs emissions and, thus, adverse environmental impacts, a TEG recovery system can potentially reduce energy consumption and reduce the harmful gases into the atmosphere. However, several essential strategies must be considered before installing a TEG system on the rotary kiln surfaces, as follows:

1. It is necessary to assess the surface temperature balance along the rotary kiln length to reduce side effects on the quality of the final products and the kiln service life (e.g., elimination of crack formation on the kiln shell due to hot spots)
2. Optimizing the distance of the TEG system to the rotary kiln
3. TEGs should not be installed directly on the kiln surface because they can increase the weight of the kiln and, thus, require more energy to rotate. Moreover, the direct installation of TEGs on the kiln surface adds a thermal resistance layer to the outer surface.
The achievements mentioned above in regard to waste heat recycling from a cement rotary kiln are just some examples of the application of thermometric generators for industrial waste heat recovery. However, many more industrial sectors, such as concentrated solar plants, steel industries, and other industrial sectors, produce heat as a by-product and can be the subject of experimental and field-test evaluations of the performance of thermoelectric heat recovery systems.

## 6. Challenges and Opportunities

As previously mentioned, TEGs can convert waste heat into electrical energy. Such capability should enable a reduction in fossil fuel consumption and decrease GHGs emissions to some extent. Compared with conventional generators using fossil fuels, the main challenge of TEGs is their low work efficiency, which is directly related to TE materials. Therefore, there are challenges in developing new and more advanced TE materials with larger zT factors. This is now the main drive in $\mathrm{R} \& \mathrm{D}$ activities in laboratories worldwide [31,32,66,92]. In addition to the development of new materials, the architecture of TEGs and their ability to adapt with the geometrical variations when TEGs come in contact with surfaces, where there is a temperature gradient, is a further challenge faced by the TE community. The selection of materials for the body of TEGs introduces some new constraints. In addition to flexibility in adapting different shapes, they have to be good thermal conductors to transfer waste heat to TE materials. Another challenge faced by the TE community is the cost of TEGs, which is attributed to the cost of currently available TE materials and the employed fabrication route, which is mainly based on powder metallurgy. The introduction of additive manufacturing and processes such as powder bed fusion using a laser (L-PBF) or electron beam (E-PBF) as the primary source of input energy or direct energy deposition, DED, could provide some solutions for reducing cost. However, despite such challenges, the advent and introduction of TEGs provide some opportunities to further advance our societies with better access to health, information and monitoring through the introduction of advanced sensors, vibration-free cooling, and heating systems, to name just a few. As mentioned before, the most pronounced opportunity for using TEGs is to reduce the consumption of fossil fuels. TEGs use waste heat to produce electrical energy rather than burning extra fossil fuels for a specific energy output or the requirement of a complex infrastructure (e.g., solar and wind power plants) for collecting the thermal energy and converting it into electricity. Furthermore, TEGs are not weather or daytimedependent, as in photo voltaic-based energy generators where solar energy is the vital ingredient. Furthermore, TEGs can be installed in various orientations with no impact on their performances.

Moreover, in some particular applications, such as NASA missions [26,27] outside the earth where there is no access to solar energy or in highly sensitive to vibration instruments, such as the Hubble telescope (e.g., in solid dewar) [100], TE devices are
the only options. Indeed, TE devices are the preferred options if reliability or accuracy matters more than efficiency. These features have opened a wide range of applications for TE devices, specifically for low-temperature applications, such as wearable TEGs and thermoelectric cooling systems in medical practices [101]. Self-powered wearable electronics are expected to become more extended and advanced from the wrist and into textiles in people's daily lives [102,103]. Given the immense research and vast commercial prospects in self-powered wearable electronics, new near-term advancements are expected. In parallel, the market for self-powered wearable electronics is likely to continue its fast developments and retain its strategy to improve and modify people's lives.

## 7. Conclusions

The current research presented an overview of the trends in energy management, emphasizing waste heat recovery by TEGs. It is claimed the new TE-based technology will produce clean energy and reduce fossil fuel consumption through recycling waste heat.

Direct advantages achieved by introducing heat recycling technology can lead to a substantial advancement in energy consumption efficiency. In addition, heat recycling technology can recover part of the waste heat and significantly reduce greenhouse gas emissions and enhance public health with better sustainability. In this approach, thermoelectric generators enable the industries to capture waste heat and recycle it by direct conversion into electrical energy. Furthermore, the simplicity of thermoelectric generators with no mechanical parts has extended their applications to a range of situations where vibration is a hurdle, such as precision positioning in medicine or photography, as in the case of the Hubble telescope. In addition, the advent of new TE materials has enabled various industries to benefit from energy recovery and recycling, where specifically heat is produced with a temperature of several hundred degrees as a by-product. In order to better highlight the requirements for the introduction of waste heat recovery and recycling, we provided a brief account of the science of TE materials to emphasize further the need for more development of advanced TE materials. Finally, as a case study, recent attempts in the successful application and performance of TEGs to recover waste heat emitted from rotary cement kilns have been discussed. However, due to the growing need of industrial sectors to employ waste heat recycling systems and save energy and costs, more experimental and field tests are still required to commercialize TEG recovery systems.

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