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500 kHz or 8.5 GHz? And all the ranges in between.





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ABSTRACT

We develop a machine-learning method for coarse-graining condensed-phase molecular systems using anisotropic particles. The method extends currently available high-dimensional neural network potentials by addressing molecular anisotropy. We demonstrate the flexibility of the method by parametrizing single-site coarse-grained models of a rigid small molecule (benzene) and a semi-flexible organic semiconductor (sexithiophene), attaining structural accuracy close to the all-atom models for both molecules at a considerably lower computational expense. The machine-learning method of constructing the coarse-grained potential is shown to be straightforward and sufficiently robust to capture anisotropic interactions and many-body effects. The method is validated through its ability to reproduce the structural properties of the small molecule's liquid phase and the phase transitions of the semi-flexible molecule over a wide temperature range.

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I. INTRODUCTION

Machine learning is quickly becoming an invaluable tool in the search, analysis, and development of new materials.^{1,2} Neural networks, in particular, have had major recent success in areas ranging from predicting the folded conformation of biological macromolecules such as proteins³ to developing highly accurate temperature-transferable interatomic potentials.^{4,5} The latter is an important advance in the field of molecular dynamics (MD) simulations. Improvements in these machine-learning models aim to expand the length and time scale of simulations without sacrificing accuracy.^{6,7} Currently used *ab initio* molecular dynamics simulation models are generally accurate but are computationally expensive, limiting their ability to probe long time scales.^{8,9} However, machine-learning potentials can produce *ab initio* accuracy at the computational cost of classical atomistic models.^{10,11}

Even though simulations at the classical MD level are faster than *ab initio* MD, the speedup is still insufficient to model the large length and time scales needed to fully understand certain phenomena and processes such as supramolecular assembly. It is well known that explicit modeling of high-frequency motion is not critical for describing many phenomena in molecular systems. These simplifications have led to the development of molecular coarsegrained models to study large, complex materials and biological systems.¹² Parameterization of coarse-grained interaction potentials commonly uses one of two strategies: the top-down approach, in which parameters are tuned to match macroscopic observables, as exemplified by the Martini model,¹³ and the bottom-up approach, in which interactions are derived from the properties of a fine-grained model with more degrees of freedom.¹² By following a similar bottom-up process used to apply machine learning to *ab initio* MD data, neural-network approaches have been extended to coarsegrained molecular models, further extending the length and time scale of simulations with atomistic accuracy.^{14,15}

Integrating out fine-grained degrees of freedom in the coarse-graining process introduces many-body interactions between coarse-grained particles, even if the underlying fine-grained model contains only pair-wise interactions. Thus, neural-network potentials using isotropic coarse-grained particles have several advantages over their pair-wise additive analytical counterparts since they are constructed as many-body potentials. This many-body potential can become costly when multiple coarse-grained particles are needed to preserve the shape anisotropy of a molecular fragment. It is sometimes more accurate and computationally efficient to represent these groups of atoms as a single anisotropic coarse-grained particle, such as an ellipsoid, as in the case of large, rigid, anisotropic molecular fragments. Analytical anisotropic coarse-grained potentials such as the Gay–Berne potential^{16,17} were developed to address the poor performance of spherically symmetric potentials in replicating intrinsic anisotropic interactions such as π -stacking. By modeling rigid anisotropic groups of atoms as ellipsoids, the anisotropic properties of the group are preserved in a single-site model. Shape and interaction anisotropy are especially important for the study of organic semiconductor molecules, which typically consist of highly anisotropic and rigid π -conjugated units and often form liquid-crystal phases whose morphology strongly affects their performance in devices such as solar cells, transistors, and light-emitting diodes.¹⁸

Unlike analytical pair-wise additive potentials such as the Gay-Berne potential, high-dimensional neural-network potentials are constructed based on the immediate neighborhood of a molecule and, thus, account for many-body effects as well as local density variations. Notable machine-learning implementations of interatomic and intermolecular potentials include the neural-network potentials developed by Behler et al.¹⁹ The Behler neural-network potentials are constructed from a set of symmetry functions used to represent the invariant properties of the atomic environment of each atom taken from *ab initio* simulations. DeepMD¹⁰ and DeepCG¹⁴ are two other neural-network codes constructed for atomistic and coarsegrained simulations, respectively. All these neural-network potentials rely on an invariant representation of the atomic/molecular environment. The CGnets deep-learning approach¹⁵ employs a prior potential to account for areas in a coarse-grained dataset that may not be properly sampled due to high repulsive energies. These interactions are especially important to reproduce the local structure of the simulated material.

Machine learning has previously been applied to the parameterization of coarse-grained models with anisotropic particles,²⁰ but no such implementation has used a nonlinear neural-network optimization method to construct the coarse-grained potential. In this work, we address this gap in knowledge by using a neural network to construct a high-dimensional anisotropic coarsegrained potential. We parameterize the neural-network potential using a recently derived systematic and general bottom-up coarsegraining method called anisotropic force-matching coarse-graining (AFM-CG),²¹ which generalizes the multi-scale coarse-graining (MS-CG) method²² for isotropic coarse-grained particles to anisotropic particles. The method rigorously accounts for finitetemperature, many-body effects without assuming a specific functional form of the anisotropic coarse-grained potential. It yields general equations relating the forces, torques, masses, and moments of inertia of the coarse-grained particles to the properties of a finegrained (e.g., all-atom) MD simulation based on a mapping between fine-grained and coarse-grained coordinates and momenta and by matching the equilibrium coarse-grained phase-space distribution with the mapped distribution of the fine-grained system. The previous implementations of the AFM-CG method approximated the coarse-grained potential as a sum of pair interactions between particles.²¹ Here, we extend this approach to more general many-body anisotropic interactions described by a neural-network potential. We also extend the approach, which was derived for constantvolume systems in the canonical ensemble to constant-pressure

systems by applying a virial-matching condition previously derived for the MS-CG method. $^{\rm 23}$

A general coarse-grained potential should capture any temperature-dependent phase transition associated with either melting, annealing, or glass transition temperatures as well as the local structure and density of the material. The focus is on the development of a model for which trained parameters can be easily obtained and one capable of reproducing interaction anisotropy, temperature transferability, and many-body effects. The flexibility of the new model is demonstrated through the matching of structural and thermodynamic properties of condensed-phase systems of a small anisotropic molecule, benzene, and of a larger, more flexible organic semiconductor molecule, sexithiophene. These two molecules were chosen to determine the conditions under which coarse-grained structural inaccuracy outweighs the computational efficiency of a single-anisotropic-site model.

II. THEORY

The key aspects of the theory that underpins the AFM-CG method and its extension to constant pressure via virial matching are summarized below. The reader is referred to Ref. 21 for a more detailed description and the full derivation of the method.

A. Fine-grained to coarse-grained mapping

The positions $\mathbf{r}^n = \mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_n$ of the *n* fine-grained particles are mapped onto the positions $\mathbf{R}^N = \mathbf{R}_1, \mathbf{R}_2, \dots, \mathbf{R}_N$ and orientations $\mathbf{\Omega}^N = \mathbf{\Omega}_1, \mathbf{\Omega}_2, \dots, \mathbf{\Omega}_N$ of the *N* anisotropic coarse-grained particles, where \mathbf{r}_i is the position of fine-grained particle *i*, and \mathbf{R}_I and $\mathbf{\Omega}_I$ are the position and orientation, respectively, of coarse-grained particle *I*, as shown in Figs. 1(a) and 1(b) for the α -sexithiophene and benzene molecules, respectively. Each fine-grained particle *i* is mapped to a single coarse-grained particle by defining *N* nonintersecting subsets, $\zeta_1, \zeta_2, \dots, \zeta_N$, of the fine-grained particle indices such that ζ_I contains the indices of the fine-grained particles mapped onto coarse-grained particle *I*. The position \mathbf{R}_I of coarsegrained particle *I* is defined to be equal to the center-of-mass of the group of fine-grained particles that are mapped onto it, i.e.,

$$\boldsymbol{R}_{I} = \frac{\sum_{i \in \zeta_{I}} m_{i} \boldsymbol{r}_{i}}{\sum_{i \in \zeta_{I}} m_{i}},$$
(1)

where m_i is the mass of fine-grained particle *i* and $\sum_{i \in \zeta_i}$ denotes a sum over the fine-grained particles in the set ζ_I of particles mapped onto the coarse-grained particle *I*. The orientation

$$\boldsymbol{\Omega}_{I} = \begin{bmatrix} \boldsymbol{\Omega}_{I,1} \\ \boldsymbol{\Omega}_{I,2} \\ \boldsymbol{\Omega}_{I,3} \end{bmatrix}$$
(2)

of coarse-grained particle *I* is specified by the rotation matrix whose components are the particle's three normalized principal axes of inertia, $\Omega_{I,q}$ for q = 1, 2, 3. These axes are defined to be equal to the corresponding principal axes relative to the center-of-mass of the

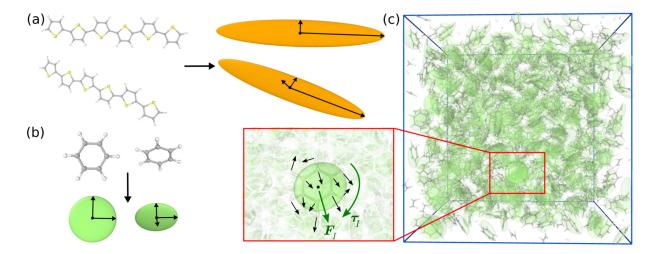


FIG. 1. Mapping of fine-grained atom positions to position and orientation coordinates of anisotropic coarse-grained particles for pairs of (a) α -sexithiophene and (b) benzene molecules, in which each molecule is mapped to a single coarse-grained particle. Carbon, hydrogen, and sulfur atoms in the all-atom structures are colored gray, white, and yellow, respectively. The arrows on the coarse-grained particles represent the principal inertia axes that define the particle orientation, which intersect at the center-of-mass defining the particle position. (c) All-atom benzene system in which the coarse-grained representation has been overlaid; the magnified image shows the atomistic forces (black arrows) that contribute to the coarse-grained force F_I and torque τ_I (green arrows) on a specific molecule.

group of fine-grained particles that are mapped onto the coarsegrained particle. Thus, these axes are the normalized eigenvectors of the inertia tensor

$$\mathbb{I}_{\mathrm{FG},I} = \sum_{i \in \zeta_I} m_i (\|\Delta \boldsymbol{r}_i\|^2 \boldsymbol{E} - \Delta \boldsymbol{r}_i \Delta \boldsymbol{r}_i^{\mathrm{T}}), \qquad (3)$$

where $\Delta \mathbf{r}_i = \mathbf{r}_i - \mathbf{R}_I$ is the position of fine-grained particle *i* relative to the center-of-mass (coarse-grained particle position), \mathbf{E} is the 3×3 identity matrix, $\| \cdots \|$ denotes the vector norm, and \mathbf{r}_i^T is the transpose of the vector \mathbf{r}_i . From these coordinate mappings and the relationship between generalized coordinates and momenta from Hamilton's equations,²⁴ mappings from the linear momenta $\mathbf{p}^n = \mathbf{p}_1, \mathbf{p}_2, \dots, \mathbf{p}_n$ of the fine-grained particles to the linear momenta $\mathbf{P}^N = \mathbf{P}_1, \mathbf{P}_2, \dots, \mathbf{P}_N$ and angular momenta $L^N = L_1, L_2, \dots, L_N$ of the anisotropic coarse-grained particles can also be defined, where \mathbf{p}_i is the momentum of fine-grained particle *i*, and \mathbf{P}_I and L_I are the linear and angular momentum, respectively, of coarse-grained particle I.²¹ The mappings for coarse-grained particle *I* are

$$\boldsymbol{P}_{I} = \frac{M_{I}}{\sum_{i \in \zeta_{I}} m_{i}} \sum_{i \in \zeta_{I}} \boldsymbol{p}_{i}$$
(4)

and

$$\boldsymbol{L}_{I} = \mathbb{I}_{I} \mathbb{I}_{\mathrm{FG},I}^{-1} \sum_{i \in \zeta_{I}} \Delta \boldsymbol{r}_{i} \times \boldsymbol{p}_{i}, \qquad (5)$$

respectively, where \mathbb{I}_I is the inertia tensor of coarse-grained particle *I*, and × denotes the vector (cross) product.

B. Configuration-space consistency conditions

Given these mappings, several conditions can be derived that the coarse-grained model must satisfy for its equilibrium coarse-grained phase-space distribution to match the corresponding mapped distribution of the fine-grained system. Consistency between the configuration-space distributions gives the following matching conditions between the forces F_I and torques τ_I on coarse-grained particle *I* and the forces on the fine-grained particles mapped onto it:²¹

$$F_{I}(\boldsymbol{R}^{N},\boldsymbol{\Omega}^{N}) = -\frac{\partial U}{\partial R_{I}} = \left\langle \sum_{i \in \zeta_{I}} f_{i} \right\rangle_{\boldsymbol{R}^{N},\boldsymbol{\Omega}^{N}}$$
(6)

and

$$\boldsymbol{\tau}_{I}(\boldsymbol{R}^{N},\boldsymbol{\Omega}^{N}) = -\sum_{q} \Omega_{I,q} \times \frac{\partial U}{\partial \Omega_{I,q}} = \left\langle \sum_{i \in \zeta_{I}} \Delta \boldsymbol{r}_{i} \times \boldsymbol{f}_{i} \right\rangle_{\boldsymbol{R}^{N},\boldsymbol{\Omega}^{N}}, \quad (7)$$

where $U(\mathbb{R}^N, \Omega^N)$ is the coarse-grained potential, $f_i(\mathbf{r}^n) = -\frac{\partial u}{\partial r_i}$ is the force on fine-grained particle *i*, with $u(\mathbf{r}^n)$ denoting the fine-grained potential and $\langle \cdots \rangle_{\mathbb{R}^N,\Omega^N}$ denoting an average over fined-grained configurations mapped to coarse-grained configuration \mathbb{R}^N , Ω^N . The fine-grained atomistic forces that contribute to the coarse-grained force and torque on a molecule in a liquid benzene system are shown in Fig. 1(c).

C. Momentum-space consistency conditions

Consistency between the momentum-space distributions requires the mass M_I of coarse-grained particle I to be the sum of the masses of its constituent fine-grained particles, i.e.,²¹

$$M_I = \sum_{i \in \zeta_I} m_i. \tag{8}$$

In addition, if it is assumed that the principal moments of inertia of the coarse-grained particles are constant (which is necessary for

the coarse-grained model to be compatible with standard molecular dynamics simulation algorithms and codes), momentum-space consistency between coarse-grained and fine-grained models requires the angular momentum (or, equivalently, angular velocity) distribution of the fine-grained system to be separable into independent factors for each coarse-grained particle and for these factors to be independent of the system configuration, which leads to the condition²¹

$$I_{l,q}^{1/2} \exp\left(-\frac{I_{I,q}\omega_{l,q}^2}{2k_{\rm B}T}\right) \approx \left(I_{\rm FG,I,q}^{1/2} \exp\left(-\frac{I_{\rm FG,I,q}\omega_{l,q}^2}{2k_{\rm B}T}\right)\right)_{R_l,\Omega_l},\qquad(9)$$

where $I_{I,q}$, $I_{\text{FG},I,q}$, and $\omega_{I,q}$ are the components of the coarse-grained moment of inertia, fine-grained moment of inertia, and angular velocity about the *q* axis, respectively and $\langle \cdots \rangle_{R_i,\Omega_i}$ denotes an equilibrium average of fine-grained configurations consistent with the coordinate mapping of coarse-grained particle *I*. The assumption of separability and configuration independence of the angular momentum distribution should be accurate for groups of atoms that behave approximately as rigid bodies. However, it is unclear if this assumption is accurate, in general, for flexible molecules. However, the choice of the inertia tensor components in the coarse-grained model does not affect any of the key results on structural and thermodynamic properties presented later on, so we do not dwell on the accuracy of this choice. Furthermore, if the fluctuations in $I_{\text{FG},I,q}$ are small compared to its mean, it can be shown that²¹

$$I_{I,q} \approx \left\langle I_{\mathrm{FG},I,q} \right\rangle_{\boldsymbol{R}_{I},\boldsymbol{\Omega}_{I}},\tag{10}$$

i.e., the principal moment of inertia of a coarse-grained particle about each principal axis q is approximately equal to the equilibrium average of the corresponding principal moment of the fine-grained particles mapped onto it.

D. Virial matching

The AFM-CG method was derived only for the constantvolume conditions of the canonical ensemble but is straightforwardly generalized to constant-pressure conditions by analogy with the MS-CG method for spherical coarse-grained particles in the isothermal-isobaric ensemble.²³ Thus, the force- and torquematching conditions at constant pressure are the same as those in Eqs. (6) and (7), except that the coarse-grained forces, torques, and potential are, in general, functions of the coarse-grained system volume *V* and the equilibrium average is constrained to configurations in which the fine-grained system volume v = V. The coarse-grained potential must also satisfy a virial-matching condition²³

$$W(\mathbf{R}^{N}, \mathbf{\Omega}^{N}, V) = -\frac{\partial U}{\partial V}$$
$$= \left\langle \frac{(n-N)k_{\rm B}T}{\nu} + \frac{1}{3\nu} \sum_{i=1}^{n} \mathbf{f}_{i} \cdot \mathbf{r}_{i} \right\rangle_{R^{N}, \mathbf{\Omega}^{N}, V}.$$
(11)

E. Summary

In summary, for the equilibrium phase-space distribution of the coarse-grained model to match that of the fine-grained model in the isothermal-isobaric ensemble, the coarse-grained potential should satisfy Eqs. (6), (7), and (11), while the coarse-grained masses and

principal moments of inertia should satisfy Eqs. (8) and (9), respectively. As shown below, using the more approximate Eq. (10) to parameterize the moments of inertia gives almost the same results as Eq. (9), even for a flexible molecule, so we have used this simpler equation for parameterization later on.

III. METHODS

A. Force-, torque-, and virial-matching algorithm

The analytical expression for the coarse-grain potential U is not usually known. However, an approximation to the functional form can be obtained using a neural-network optimization algorithm with Eqs. (6), (7), and (11) acting as necessary constraints. In general, $U(\mathbf{R}^N, \mathbf{\Omega}^N, V)$ is a function of the particle configuration and system volume. In this work, we have assumed that U does not depend explicitly on V, in which case²³

$$\frac{\partial U}{\partial V} = \frac{1}{3V} \sum_{I=1}^{N} \frac{\partial U}{\partial \boldsymbol{R}_{I}} \cdot \boldsymbol{R}_{I}.$$
(12)

With this approximation, the virial-matching condition in Eq. (11) can be written, using v = V, as

$$-\sum_{I=1}^{N} \frac{\partial U}{\partial \boldsymbol{R}_{I}} \cdot \boldsymbol{R}_{I} = \left(3(n-N)k_{\mathrm{B}}T + \sum_{i=1}^{n}\boldsymbol{f}_{i} \cdot \boldsymbol{r}_{i}\right)_{\boldsymbol{R}^{N},\boldsymbol{\Omega}^{N},\boldsymbol{V}}.$$
 (13)

Despite this approximation, we show later on that the coarse-grained models parameterized accurately match the average density of the corresponding all-atom fine-grained system at constant pressure. Note that previous neural-network coarse-graining implementations for isotropic particles^{14,15} have not used pressure-matching in neural-network training and were restricted to constant-volume systems in which the density was not allowed to vary, so the ability of the resulting models to describe the system density accurately at constant pressure is not possible to ascertain.

To ensure that all equivalent configurations are assigned the same position in coordinate space, a transformation was made from the set of Cartesian coordinates to a vector D_{IJ} that was invariant under translation, rotation, and permutation of any pair of coarsegrained particles *I* and *J*,^{10,25–27} which was defined in terms of the positions, R_I and R_J , and orientations, Ω_I and Ω_J , of the two particles by

$$D_{IJ} = \{ R_{IJ}, R_{IJ} \cdot \Omega_{I,1}, R_{IJ} \cdot \Omega_{I,2}, R_{IJ} \cdot \Omega_{I,3}, R_{IJ} \cdot \Omega_{J,1}, R_{IJ} \cdot \Omega_{J,2}, R_{IJ} \cdot \Omega_{J,3}, \Omega_{I,1} \cdot \Omega_{J,1}, \Omega_{I,1} \cdot \Omega_{J,2}, \Omega_{I,1} \cdot \Omega_{J,3}, \Omega_{I,2} \cdot \Omega_{J,1}, \Omega_{I,2} \cdot \Omega_{J,2}, \Omega_{I,2} \cdot \Omega_{J,3}, \Omega_{I,3} \cdot \Omega_{I,1}, \Omega_{I,3} \cdot \Omega_{I,2}, \Omega_{I,3} \cdot \Omega_{I,3} \},$$
(14)

where $R_{IJ} \equiv ||\mathbf{R}_{IJ}||$, $\mathbf{R}_{IJ} \equiv \mathbf{R}_I - \mathbf{R}_J$, and Ω_I and Ω_J are specified by rotation matrices of the form of Eq. (2). The coordinates of each neighbor within the cut-off distance of particle *I* were transformed to a D_{IJ} vector. All the D_{IJ} vectors for a given neighborhood were concatenated into a 2D matrix \mathbb{D}_I of size $N \times \dim(D_{IJ})$ representing a unique configurational fingerprint for coarse-grained particle *I*.

The potential function could then be written in terms of a set of neural network trainable parameters and activation functions transforming \mathbb{D}_I to a potential energy value. Although, \mathbb{D}_I is a sufficient

specification of the coarse-grained coordinates to enforce the relevant invariant properties of the molecular environment, it does not possess all the symmetries of the potential energy surface that it aims to fit.^{25,28} For each molecular environment, it was assumed that the interactions were predominantly short-ranged such that neighbors beyond a certain cut-off distance, R_c , did not contribute to the potential.¹⁹ This condition can be enforced by a cut-off function of the form

$$g_{c}(R_{IJ}) = \begin{cases} \frac{1}{2} \left[\cos\left(\frac{\pi R_{IJ}}{R_{c}}\right) + 1 \right], & R_{IJ} \leq R_{c}, \\ 0, & R_{IJ} > R_{c}. \end{cases}$$
(15)

A set of these cut-off functions can enforce the radial symmetry conditions of the underlying potential energy surface by storing information about the radial distribution of neighbors, using¹⁹

$$G_{I}^{1} = \sum_{J \neq I} g_{c}(R_{IJ}).$$
(16)

Continuity of the potential along angular dimensions was ensured by using a compression layer to learn a set of collective variables from vector D_{IJ} , which were constrained by a modification of the well-behaved G^5 symmetry function¹⁹ that describes the relative orientation of anisotropic particles instead of the angle between triplets of isotropic particles and is given by

$$G_{I}^{5} = \sum_{J \neq I} \prod_{\mu=1}^{M} 2^{1-\nu} (1 + \lambda \cos \theta_{IJ,\mu})^{\nu} e^{-\eta (R_{IJ} - R_{s})^{2}} g_{c}(R_{IJ}), \qquad (17)$$

where $\lambda \in \{-1, 1\}$ and R_s , ν , and η are tunable hyperparameters, while $\{\cos \theta_{IJ,\mu}\}$ is the set of machine-learned collective variables with the same properties as the angular component of the underlying potential, and M is the total number of machine-learned angular variables. These angular symmetry functions store information about the angular-radial distribution of neighbors in the local environment of coarse-grained particle I. Unlike the case of spherically symmetric particles, in a local reference frame, a neighboring anisotropic particle requires a minimum of seven independent scalar variables to fully describe its position and orientation. However, previous implementations of analytical potentials, including the Gay-Berne potential,^{16,17} have used fewer coordinates for the calculation of the potential and forces. Similarly, for the neural-network potential, an additional compression layer was included to remove the redundant angles from the D_{II} vectors since the combination of translation and rotation in 3D is parameterized by at most seven unique coordinates. The Behler symmetry functions were enforced on the output of the compression layer, ensuring that the learned compression had the same symmetry and continuity as the underlying potential. The reduction in the dimension of D_{II} is expected to lead to improved generalization and reduced over-fitting on a small dataset, which can result in more efficient training of an accurate potential.

For each CG particle *I* with a nearest-neighbor molecular environment represented by the set of distances R_{IJ} and angles $\cos \theta_{IJ,\mu}$, a set of these symmetry functions with tuned hyperparameters $(\lambda, \nu, \eta, R_s, \text{and } R_c)$ was used as the input to a dense neural network to predict the potential energy of the molecular environment. Symmetry functions used to represent the local environment were constructed using all possible permutations of values from a specified set of hyperparameters. The neural network initially used eight symmetry functions, including a single G^1 symmetry function parameterized by the initial guess of the optimal cut-off radius. New symmetry functions were added progressively to the set if they resulted in a significant reduction in the neural-network cross-validation loss, which is defined below, compared with the preceding set of symmetry functions. The set of hyperparameters in the symmetry functions used in the anisotropic coarse-grained models parameterized in this work can be found in the supplementary material.

To further reduce the amount of data needed to train the neural network, a prior repulsive potential was defined with pairwise additive properties. This potential was used to ensure physical behavior in the regions of the potential where the forces are large and, thus, are rarely sampled in an equilibrium MD simulation. This prior potential only needs to satisfy two conditions: first, it must be repulsive at short radial separations, and, second, the position of its repulsive barrier must be orientationally dependent. A simple equation satisfying these conditions is

$$U_{\text{prior},I} = \sum_{J \neq I} B_1 \sigma_c (\mathbb{D}_I)^{-B_2}, \qquad (18)$$

where σ_c is a neural-network compression layer function, and B_1 and B_2 are strictly positive trainable parameters. It is also possible to achieve a similar large repulsive barrier through a more advanced nonlinear sampling of the MD simulation data. The prior potential fits the purely repulsive part of the angular-dependent potential to the molecular environment, while the neural-network potential fits the attractive and oscillatory corrections to the environment. The final prediction for the potential energy of the environment of coarse-grained particle *I* is, therefore, the sum of the neural-network potential $U_{NN,I}$ and the prior repulsive potential U_{prior,I_3} ¹⁵

$$U_I = U_{\text{NN},I} + U_{\text{prior},I}; \tag{19}$$

thus, the total coarse-grained potential is

$$U = \sum_{I=1}^{N} U_{I}.$$
 (20)

From the matching conditions in Eqs. (6), (7), and (13), optimization of the neural-network weights and biases requires a loss function of the form

$$L = \left\{ \sum_{I=1}^{N} \left(\alpha \left| \boldsymbol{F}_{\mathrm{FG},I} + \frac{\partial U}{\partial R_{I}} \right|^{2} + \beta \left| \boldsymbol{\tau}_{\mathrm{FG},I} + \sum_{q} \Omega_{I,q} \times \frac{\partial U}{\partial \Omega_{I,q}} \right|^{2} \right) + \gamma \left| 3(n-N)k_{\mathrm{B}}T + \sum_{I=1}^{N} \left(\bar{W}_{\mathrm{FG},I} + \frac{\partial U}{\partial \boldsymbol{R}_{I}} \cdot \boldsymbol{R}_{I} \right) \right|^{2} \right\}_{R^{N},\Omega^{N},V}, \quad (21)$$

where

$$\boldsymbol{F}_{\mathrm{FG},I} \equiv \sum_{i \in \zeta_{I}} \boldsymbol{f}_{i}, \ \boldsymbol{\tau}_{\mathrm{FG},I} \equiv \sum_{i \in \zeta_{I}} \Delta \boldsymbol{r}_{i} \times \boldsymbol{f}_{i}, \ \bar{W}_{\mathrm{FG},I} \equiv \sum_{i \in \zeta_{I}} \boldsymbol{f}_{i} \cdot \boldsymbol{r}_{i},$$
(22)

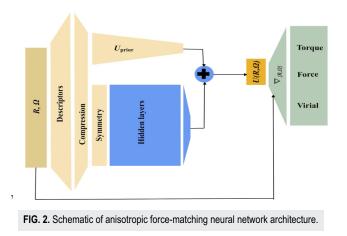
and α , β , and γ are weights. These weights specify the fraction of each loss that is used for backpropagation and were free to change

with the learning rate during optimization.¹⁴ Even though there have been significant efforts to develop methods to fit the constrained averaged coarse-grained forces directly,^{29,30} the average total fine-grained forces subject to the constraint of matching fine-grained and coarse-grained configurations are not easily obtained. An indirect means of minimizing the loss function in Eq. (21) is possible by replacing the constrained ensemble average with an average over instantaneous unconstrained simulation configurations¹⁴

$$L_{\text{inst}} = \frac{1}{N_{\text{t}}} \sum_{l=1}^{N_{\text{t}}} \left[\sum_{I=1}^{N} \left(\alpha \left| \boldsymbol{F}_{\text{FG},I}(\boldsymbol{r}_{t}^{n}) + \frac{\partial U(\boldsymbol{\xi}_{t})}{\partial \boldsymbol{R}_{I}} \right|^{2} + \beta \left| \boldsymbol{\tau}_{\text{FG},I}(\boldsymbol{r}_{t}^{n}) + \sum_{q} \Omega_{I,q}(\boldsymbol{\xi}_{t}) \right) \times \frac{\partial U(\boldsymbol{\xi}_{t})}{\partial \Omega_{I,q}} \right|^{2} \right] + \gamma \left| 3(n-N)k_{\text{B}}T + \sum_{I=1}^{N} \left(\bar{W}_{\text{FG},I}(\boldsymbol{r}_{t}^{n}) + \frac{\partial U(\boldsymbol{\xi}_{t})}{\partial \boldsymbol{R}_{I}} \cdot \boldsymbol{R}_{I}(\boldsymbol{\xi}_{t}) \right) \right|^{2} \right],$$
(23)

since it can be shown, for a sufficiently large dataset that comprehensively samples the equilibrium ensemble of the fine-grained system, that *L* and *L*_{inst} have the same global minimum. Here, *N*_t is the number of simulation configurations in the dataset, \mathbf{r}_t^n and \mathbf{v}_t are the fine-grained coordinates and system volume for configuration *t*, and $\boldsymbol{\xi}_t = (\mathbf{R}^N(\mathbf{r}_t^n), \mathbf{\Omega}^N(\mathbf{r}_t^n), \mathbf{V}(v_t))$ is the mapped coarse-grained configuration for this fine-grained configuration. The loss function was optimized using the minibatch gradient descent as implemented in TensorFlow.

The feedforward neural network shown in Fig. 2 was then trained, where the forward propagation used matrix \mathbb{D}_I as an input to predict the coarse-grained potential U, after which TensorFlow's computational derivative was used to calculate the outputs, namely the predicted forces, torques, and virial. In the backpropagation stage, the loss function was used to calculate the error between the true and predicted values, which was then used to update the network weights and biases. The errors between the true and predicted parameters were calculated using TensorFlow's mean squared error, and gradient descent was implemented using TensorFlow's Adam optimizer.³¹ Once the error of the neural network was minimized, the neural-network model was used to predict the forces, torques,



and virial. However, removing the output and derivative layers gives access to the predicted potential of mean force. By optimizing the partial derivatives of the potential instead of the potential itself, by the nature of the derivative, there will be less oscillation in the potential at the edges of the dataset close to the cut-off distances.

B. LAMMPS modification and neural network implementation

The neural network was constructed in Tensorflow (version 2.3.0)³² using the Keras (version 2.4.3) functional API³³ and saved using the Tensorflow SavedModel format. The trained neural network was implemented in LAMMPS using the Tensorflow C API and cppflow wrapper. All simulations were carried out using the LAMMPS MD software package (version 20 Nov 19).³⁴⁻³⁶ The Optimized Potentials for Liquid Simulations-All Atom (OPLS-AA) force field^{37–40} was used for all all-atom simulations with a cut-off distance of 10 Å for short-ranged non-bonded interactions. The parameters for benzene were identical to those in Ref. 21. Bond, angle, dihedral, and Lennard-Jones OPLS-AA parameters with 1.14* CM1A-LBCC partial atomic charges⁴¹ for sexithiophene were obtained from the LigParGen web server.⁴² Long-ranged electrostatic interactions were calculated with the particle-particle particle-mesh (PPPM) method^{36,43} The bonds that include hydrogen were constrained using the SHAKE algorithm.⁴⁴ Simulations were carried out in the isothermal-isobaric (NPT) ensemble at a pressure of 1 atm, with the temperature and pressure controlled by using a Nosé-Hoover thermostat and barostat.45,46

Neural-network training was carried out using data from a 25 ns all-atom simulation in which simulation configurations, forces, and velocities were saved at 2 ps intervals. The simulation snapshots from the last 20 ns were shuffled and then divided into four groups of equal size, $\{g_0, g_1, g_2, \text{and } g_3\}$. The neural network was initially trained on g_0 , while g_3 was used as the validation set. New snapshots were added from g_1 and g_2 if the mean errors of their predicted forces and torques were larger than those of the validation set. This iterative procedure increases the number of snapshots in the training set that are most dissimilar/uncorrelated to those in the validation set. The accuracy of the trained neural network was then compared to the expected accuracy determined from k-fold cross-validation.^{47,48} During k-fold cross-validation, the last 20 ns of simulation data was shuffled and divided into ten folds, $\{\psi_0, \ldots, \psi_9\}$. The model was validated on ψ_i and trained on $\bigcup_{j \neq i} \psi_j$ for all $i, j \in \{0 - 9\}$. The loss of the iterative training method was found to be identical to the k-fold cross-validation loss, which suggests that there was no significant data leakage between the training and validation sets.

The coarse-grained simulations were done using a modified version of LAMMPS in which the trained neural network was introduced to calculate the forces and energies. The dimensions of the coarse-grained sites used in the simulations were derived from the inertia tensor of the all-atom model. As a rigorous test of the ability of the coarse-grained model to capture the properties of the all-atom model under a variety of conditions in addition to the single temperature at which the neural network was trained, the equilibrium properties of equivalent coarse-grained and all-atom systems were compared in independent simulations at several different temperatures. In all cases, the total length of the coarse-grained simulation was 25 ns, with the last 20 ns used to calculate structural and thermodynamic properties. Statistical uncertainties were estimated using block averages⁴⁹ and are given as two standard errors.

IV. RESULTS AND DISCUSSION

To demonstrate the flexibility of the method, we have used our neural-network model to construct coarse-grained interaction potentials for benzene, an archetypal anisotropic small molecule, and α -sexithiophene, an organic semiconductor with significant applications in organic electronic devices.^{50–52} These molecules were selected to demonstrate the neural network's ability to handle anisotropic molecules of varying complexity, flexibility, and aspect ratio while still reproducing their structural and phase behavior.

The shape of a coarse-grained particle obtained from the anisotropic coarse-graining method is determined by the "average" shape of the fine-grained molecule or molecular fragment that is mapped to it under the parameterization conditions. Thus, the variation of the aspect ratio of the molecule or molecular fragment with temperature in the all-atom simulations can potentially be a qualitative indicator of the temperature transferability of the coarse-grained model. Here, the aspect ratio of the molecule was calculated as the ratio of the length to the breadth of the molecule, where the length was defined as the longest principal axis and the breadth was defined as the sum of the remaining two semi-axes. Unlike benzene, the thiophene–thiophene dihedral angles also have a temperature-dependent effect on the aspect ratio of sexithiophene.

Neural networks, in general, are very good at interpolation but struggle with extrapolation.^{53–56} The accuracy of the model is, therefore, expected to decrease as the aspect ratio of the molecule deviates from that at the parameterization temperature, as well as when the density distribution is sufficiently different from the parameterization temperature. By parameterizing the systems in the liquid phase, the model can capture a wider variety of fluctuations in the density of the system and the dimensions of the molecules. The average size of a flexible molecule in the isotropic phase will be different from the size of the molecule when locked in a rigid crystal structure.^{57,58} However, this temperature-dependent size difference should decrease with increasing rigidity of the molecule.

A. Benzene

Simulations consisting of 500 benzene molecules were carried out at 280, 300, 320, 330, and 350 K, and the coarse-grained neuralnetwork model was parameterized at 300 K. The time step was 2 fs in the all-atom simulations and 12 fs in the coarse-grained simulations. The cut-off distance hyperparameter R_c was 10 Å. The root mean squared validation error per molecule for the forces was 2.55 kcal mol⁻¹ Å⁻¹ and that for the torque was 4.35 kcal mol⁻¹. The root mean squared validation error in the pressure was 0.0092 atm.

The average principal moments of inertia in the all-atom simulation at 300 K were used to determine the principal moments of the coarse-grained model using Eq. (10) (values given in the supplementary material) since the fluctuations in the moments at the parameterization temperature were small.²¹ The variation of the molecular aspect ratio of the all-atom benzene model with temperature is shown in Fig. 3. The distribution of possible dimensions observed for benzene is narrow and remains fairly constant with



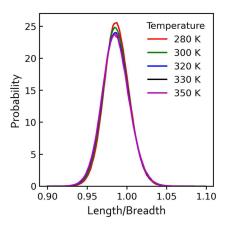


FIG. 3. Length-to-breadth ratio of the all-atom benzene model at 1 atm and various temperatures.

temperature, making benzene an ideal case where molecular flexibility does not contribute significantly to the overall error of the model. 59

Figure 4 shows that the coarse-grained neural-network model accurately captures the liquid density of the all-atom model over a wide range of temperatures, from just above the freezing point to just below the boiling point, with only slight deviations for the temperature furthest from the parameterization temperature. As shown in Fig. 5, the coarse-grained model also accurately predicts the radial distribution function (RDF) of the all-atom model over the same temperature range.

To further elucidate the accuracy of the neural-network coarsegrained model, the angular-radial distribution function (ARDF) was analyzed. The ARDF is defined by

$$g(r,\theta) = \frac{\langle n(r,\theta) \rangle}{\frac{4}{3}\pi\rho[(r+\Delta r)^3 - r^3]\sin\theta\Delta\theta},$$
(24)

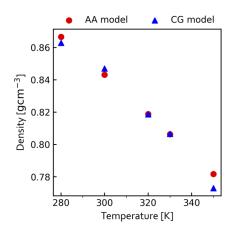


FIG. 4. Density vs temperature of the all-atom (AA) and coarse-grained (CG) benzene models at 1 atm. Error bars are smaller than the symbols (on the order of $\pm 10^{-4}$ and $\pm 10^{-3}$ g cm⁻³, respectively, for the AA and CG models).

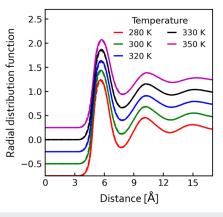


FIG. 5. Radial distribution function (RDF) of the all-atom (solid lines) and coarsegrained (dashed lines) benzene models at 1 atm and various temperatures. The RDFs have been shifted vertically for clarity.

where $\langle n(r, \theta) \rangle$ is the average number of molecules in the spherical shell within the bounds r to $r + \Delta r$ of the center-of-mass of a chosen molecule and having an out-of-plane axis rotation of θ with respect to the out-of-plane axis of the chosen molecule,⁶⁰ and ρ is the bulk number density. Figure 6 shows the 2D heatmap of the ARDF along with 1D slices of this function at specific angles at 300 K (the parameterization temperature) for the all-atom and coarse-grained models. The ARDFs at the other simulated temperatures are compared in the supplementary material. At all simulated temperatures between 280 and 350 K, the coarse-grained model captures all the major features of the fine-grain structure of the fluid. The only difference is a slight underestimation of the peak heights by the coarse-grained model. The neural-network model can demonstrate temperature transferability through the careful selection of the neural network hyperparameters to prevent the overfitting of the local number density variations.

The coarse-grained simulation of anisotropic molecules using a neural-network potential is more suited for large, preferably rigid, molecules, for which a high degree of coarse-graining can be achieved with reasonable accuracy. However, the model was still able to achieve a modest 20× speedup compared with the atomistic simulations through a combination of reduced computation time per timestep and a larger timestep. This poor performance for a small molecule such as benzene is due to the small reduction in the number of degrees of freedom from the all-atom model to the coarse-grained model, coupled with a neural-network potential that is more computationally expensive than an analytical potential. Nevertheless, computational savings are obtained even in this suboptimal case. Simulations were carried out on a 4-core Intel i7-4790K CPU, but further speedups could be achieved by taking advantage of the GPU-enabled version of TensorFlow.

B. Sexithiophene

Simulations of 512 sexithiophene molecules were carried out at 570, 590, 640, and 680 K temperatures, corresponding to temperatures previously identified in all-atom MD simulations to correspond to crystalline (K), smectic-A (SmA), nematic (N), and isotropic (I) phases, respectively.⁶¹ The time step was 1 fs in the all-atom simulations and 12 fs in the CG simulations. Although we have used the OPLS-AA force field for our all-atom simulations, whereas these previous MD simulations⁶¹ used the related AMBER force field,⁶² the structural properties of systems simulated with these two force fields (in particular, the density, orientational order parameter, and radial distribution function discussed below) are very similar for the temperature range studied. The cut-off distance hyperparameter R_c was set to 21 Å. The neural-network model was parameterized using simulation snapshots from the isotropic phase at 680 K, where the molecular mobility was highest. The conditions of the isotropic bulk phase are advantageous in efficiently sampling the configuration space, especially rare high-energy configurations

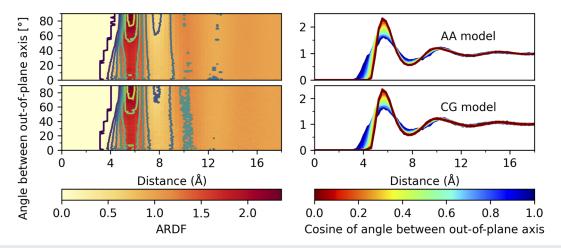


FIG. 6. Angular-radial distribution function (ARDF) of the all-atom (AA) (top) and coarse-grained (CG) (bottom) benzene models at 300 K and 1 atm depicted as a heat map (left) and 1D slices at a constant angle (right). Face-to-face, edge-to-edge, and parallel displaced configurations occur when the angle is 0°, while edge-to-face configurations occur at 90°.

necessary for the accurate reproduction of the repulsive part of the coarse-grained potential.

As shown in Fig. 7(a), the distributions of the principal moments of inertia of sexithiophene in the all-atom simulation at the parameterization temperature are broad, indicating that Eq. (10) may not be adequate for parameterizing the moments of inertia of the coarse-grained model. However, we found that using the more general Eq. (9) to parameterize the coarse-grained moments of inertia [by fitting the distributions in Figs. 7(b)–7(d)] gave values within <1%. So we used the values from Eq. (10) in the coarse-grained model.

The root mean squared validation error per molecule for the sexithiophene forces was 3.95 kcal mol⁻¹ Å⁻¹ and that for the torque was 9.8 kcal mol⁻¹. The sexithiophene final force and torque losses were larger than those of benzene because the model was

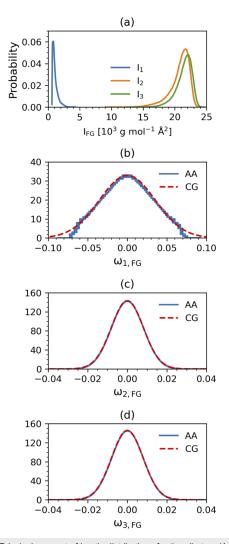


FIG. 7. (a) Principal moment of inertia distributions for the all-atom (AA) sexithiophene model at 680 K and 1 atm. The corresponding angular velocity distribution of each principal axis along with the coarse-grained (CG) fit to the distribution given by Eq. (9) is shown in (b)–(d).

not complex enough to account for the bending of the molecule and the rotation of the individual thiophene rings. The loss is also skewed toward larger values when compared with benzene because sexithiophene is a larger molecule, and so the interactions between molecules are stronger overall.

The structural properties of the coarse-grained model were compared with those of its all-atom counterpart at each of the simulated temperatures. The nonlinear change in density with respect to temperature is associated with the phase changes that occur at the simulated temperatures (Fig. 8).⁶¹ The density of the coarsegrained system agrees well with that of the all-atom system, with minimal deviations from the fine-grained system with increasing distance from the parameterization temperature. Compared with benzene, sexithiophene has a much larger change in density between the crystalline and isotropic phases. This difference results in less overlap between the local density variations in the crystalline phase at the lowest temperature and the training dataset in the isotropic phase at the highest temperature. The sexithiophene molecule is also much more flexible than benzene, as seen in the wide distribution of the aspect ratio in the all-atom model at all the simulated temperatures shown in Fig. 9, and its dimensions change significantly with temperature over the range studied. This variation of the all-atom molecular dimensions will lead to discrepancies between the angular-momentum distributions of the coarse-grained and fine-grained sexithiophene models in Eq. (9) as the temperature deviates from the parameterization temperature due to the fixed principal moments of inertia of the coarsegrained model. However, these discrepancies do not affect any of the structural or thermodynamic properties presented below. Another limitation of representing sexithiophene as a single-site ellipsoid is the loss of thiophene-thiophene torsional information. That is, for any given position and orientation of the coarse-grained ellipsoid, there are multiple different relative orientations between the thiophene groups.⁶⁵ This loss of information is significant because the anisotropic interactions of the thiophene subunits are lost, which reduces the neural network's ability to isolate which of the two short axes corresponds to the π -stacking direction.

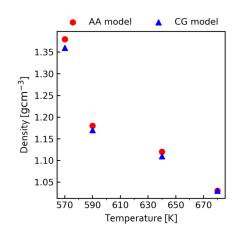


FIG. 8. Density vs temperature of the all-atom (AA) and coarse-grained (CG) sexithiophene models at 1 atm. Error bars are smaller than the symbols (on the order of $\pm 10^{-3}$ and $\pm 10^{-2}$ g cm⁻³, respectively, for the AA and CG models).

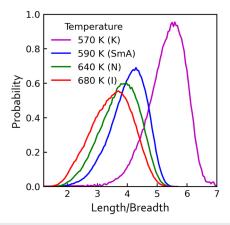


FIG. 9. Length-to-breadth ratio of the all-atom sexithiophene model at 1 atm and various temperatures. The simulated phase is given in parentheses after each temperature in the legend (I = isotropic, N = nematic, SmA = smectic A, and K = crystal).

To further confirm that the density changes were associated with transitions from the crystalline phase through the nematic and smectic phases to the isotropic phase, the scalar orientational order parameter P_2 was introduced. For a given simulation snapshot at time *t*, P_2 can be found by diagonalizing the ordering matrix

$$\boldsymbol{Q} = \frac{1}{2N} \sum_{I=1}^{N} (3\boldsymbol{u}_{I} \otimes \boldsymbol{u}_{I} - \boldsymbol{E}), \qquad (25)$$

where u_I is the unit vector along the molecular axis, and E is the identity matrix. $\langle P_2 \rangle$ is the average over the largest eigenvalue of this matrix for all snapshots of equilibrium configurations.⁶¹ Larger values of the scalar orientational order parameter close to one indicate an ordered crystalline structure, while values close to zero correspond to an isotropic disordered phase. The coarse-grained model reproduces the orientational order parameter of the all-atom model reasonably well over the temperature range simulated, as shown in Fig. 10. The coarse-grained model underestimates the degree of orientational ordering observed in the all-atom model away from the parameterization temperature, likely because it does not capture the increasing molecular shape anisotropy that is observed in the all-atom model as the temperature decreases (Fig. 9). As expected, the largest difference occurs in the predicted crystalline phase.

The same trend is seen in the radial distribution functions shown in Fig. 11, in which the agreement between the coarse-grained and all-atom models at most temperatures is excellent, with the largest deviations for the crystalline phase. The underestimation and broadening of the peaks in the crystalline radial distribution function explain the discrepancy between the order parameter of the all-atom and coarse-grained models. The observed differences are most likely due to the effect on molecular packing of the aforementioned discrepancy in molecular shape between the two models as temperature decreases.⁶⁶ Nevertheless, even in the crystalline phase, the coarse-grained model captures the peak positions of the radial distribution function very well. We are not aware of any previously developed systematically coarse-grained anisotropic model

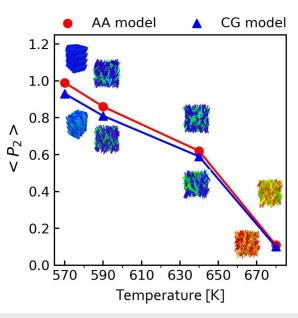


FIG. 10. Orientational order parameter vs temperature for the all-atom (AA) and coarse-grained (CG) sexithiophene models at 1 atm. Typical simulation configurations are shown at each temperature for each system (AA model above the data points and CG model below), in which the molecules have been colored according to their orientation with respect to the phase director (blue = parallel, red = perpendicular). Error bars are smaller than the symbols (on the order of $\pm 10^{-3}$ and $\pm 10^{-4}$, respectively, for the AA and CG models).

that quantitatively captures the liquid-crystal phase behavior of the all-atom model from which it was derived.

The coarse-grained model also accurately describes orientational correlations in condensed-phase sexithiophene, as illustrated by a comparison with the angular-radial distribution function of the all-atom model. At the parameterization temperature, the coarsegrained model is able to capture all major features when compared to

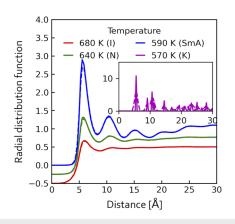


FIG. 11. Radial distribution function (RDF) of the all-atom (solid lines) and coarsegrained (dashed lines) sexithiophene models at 1 atm and various temperatures. The RDFs have been shifted vertically for clarity. The simulated phase is given in parentheses after each temperature in the legend (I = isotropic, N = nematic, SmA = smectic A, and K = crystal).

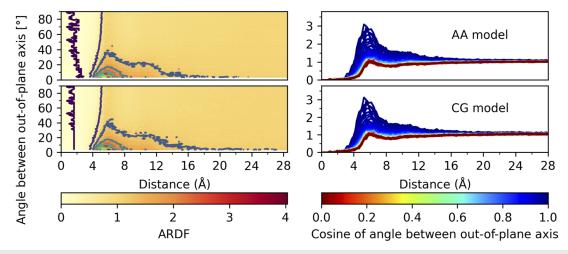


FIG. 12. Angular-radial distribution function (ARDF) of the all-atom (AA) (top) and coarse-grained (CG) (bottom) sexithiophene models at 680 K and 1 atm (isotropic phase) depicted as a heat map (left) and 1D slices at constant angle (right). Face-to-face, edge-to-edge, and parallel displaced configurations occur when the angle is 0°, while edge-to-face configurations occur at 90°.

the all-atom model (Fig. 12). The neural-network model is also able to capture the relevant features in the structure of sexithiophene's smectic liquid-crystal phase at 590 K, as shown in Fig. 13. The discrepancies in the width and height of the peaks are likely due to the differences in molecular shape away from the parameterization temperature that was mentioned earlier. The ARDFs of the two models in the nematic phase at 640 K are compared in the supplementary material and show similarly good agreement.

Despite sexithiophene not strictly meeting the conditions to be coarse-grained to a single anisotropic particle due to its significant flexibility, the coarse-grained neural-network model is still able to reproduce its condensed-phase structural properties and phase behavior with remarkable accuracy. The limitation of the singlesite model is only evident under conditions where the conformation of the molecule is highly temperature-dependent. One way to construct a neural network model that is independent of temperature would be to extract the training data from multiple temperatures and define the molecular dimensions as the average over the crystalline and isotropic phases. Although, the results for sexithiophene are substantially better than expected given its flexibility, improvements can be made to the model by considering a coarse-grained mapping consisting of more than one site.⁶⁷

The coarse-grained simulation of sexithiophene demonstrated a speed-up of 132× compared with the all-atom simulation using the

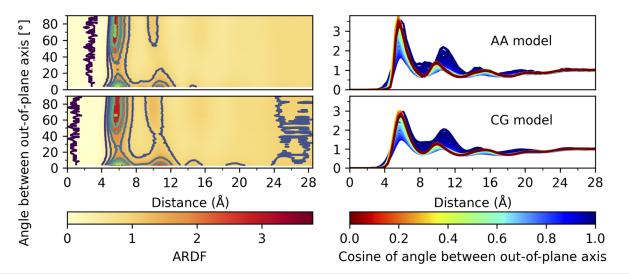


FIG. 13. Angular-radial distribution function (ARDF) of the all-atom (AA) (top) and coarse-grained (CG) (bottom) sexithiophene models at 590 K and 1 atm (smectic phase) depicted as a heat map (left) and 1D slices at constant angle (right). Face-to-face, edge-to-edge, and parallel displaced configurations occur when the angle is 0°, while edge-to-face configurations occur at 90°.

same hardware employed for the benzene simulations. This speedup is primarily due to the large reduction in the number of degrees of freedom in coarse-graining this molecule.

V. CONCLUSIONS

We have applied machine learning and a recently derived systematic coarse-graining method for anisotropic particles to develop a single-site anisotropic coarse-grained potential of a molecular system. The iterative training of the neural-network potential is able to reproduce the forces, torques, and pressure of the fine-grained all-atom system. The final loss of the iterative training model was identical to the loss obtained from k-fold cross-validation. The CG model performs well for a rigid molecule like benzene, but, remarkably, it also describes the phase behavior and molecularscale structural correlations of a flexible molecule like sexithiophene with comparable accuracy, even though the aspect ratio of the molecule changes significantly over the simulated temperature range. We have demonstrated the versatility of the coarse-graining method by parameterizing models of benzene and sexithiophene at a single temperature and then studying their accuracy in capturing the structural properties of the corresponding all-atom model at different temperatures. The sexithiophene model was also used to show the ability of the model to reproduce the phase behavior of the all-atom model, with the lowest fidelity coming from the crystalline phase, where the aspect ratio of the molecule had the largest deviation from the parameterization dataset. A natural extension to this work would be to generalize the method to a multisite anisotropic coarse-grained model for flexible molecules and polymers.

SUPPLEMENTARY MATERIAL

The supplementary material provides more extensive structural comparisons between the all-atom and coarse-grained simulations, more details on the methods used for the calculations, quantification of the agreement between structural and thermodynamic properties of the all-atom and coarse-grained models, and a list of the software needed to implement and train the neural-network potential, along with links to their GitHub repositories.

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AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Marltan O. Wilson: Formal analysis (lead); Investigation (lead); Methodology (lead); Visualization (lead); Writing – original draft (lead). David M. Huang: Conceptualization (lead); Methodology (supporting); Supervision (lead); Writing – review & editing (lead).

DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author upon reasonable request. Simulation files and neural network scripts can be found at https://doi.org/10.25909/21218057.v1.

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